

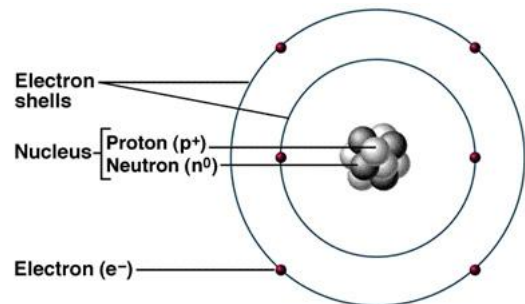
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ATOMS & THEIR COMPOSITION

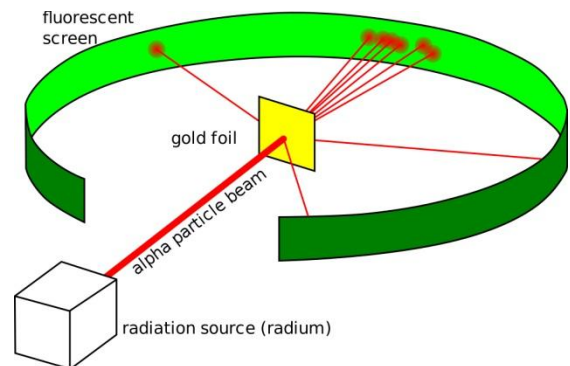
DEFINITIONS:

- ① elements =
- ② atoms =
- ③ subatomic particles =

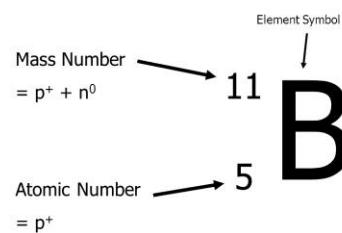
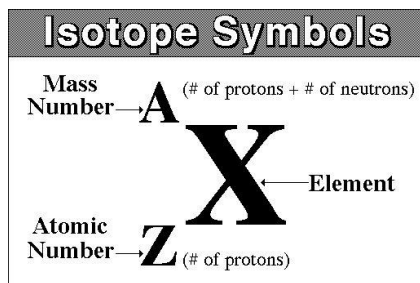
STRUCTURE OF THE ATOM:



RUTHERFORD'S GOLD FOIL EXPERIMENT:



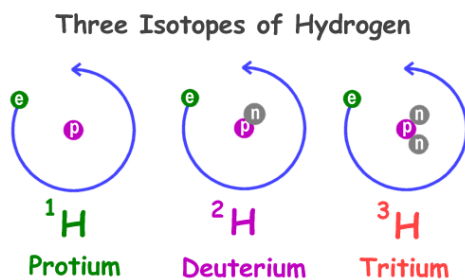
ATOMIC SYMBOLS:



DEFINITIONS:

④ isotopes =

Example:



- All isotopes of an element have similar chemical properties since they have the same number of protons & same number of electrons.

⑤ radioisotopes =

Complete the following chart (the first row is done for you):

ELEMENT	SYMBOL	Z	A	# p ⁺	# n ⁰	# e ⁻
carbon – 13	$^{13}_6\text{C}$	6	13	6	7	6
oxygen – 18			18			
		38			50	
				17	18	
iron –					29	
			33			16
					118	79
	$^{133}_{55}\text{Cs}$		133			